

SYNTHESIS AND CHARACTERISTICS OF SOME NEW ANTIMICROBIAL HETEROCYCLES COMPOUNDS OF IMIDAZOLE DERIVATIVES

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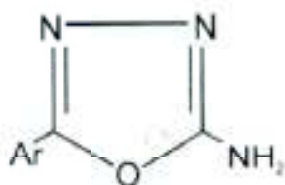
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ABSTRACT

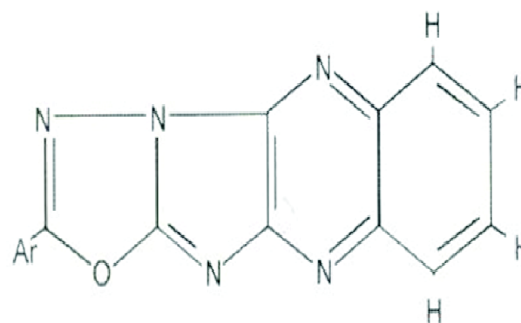
This paper describes the biological properties associated with the imidazole nucleus, it is desirable to obtain facile methods for the synthesis of fused ring systems, derived from fusion of the imidazole nucleus with other biolabile heterocyclics, because these have been less extensively studied as compared to monocyclic imidazole derivatives.

KEYWORDS: Oxadiazole, Activities, Imidazole, Heterocyclic

1,3,4-Oxadiazole ring (1) has been reported to display bactericidal, fungicidal, analgesic, antipyretic and insecticidal activities. Imidazole compounds exhibit antinflammatory (Zhang *et al.*, 2014), hypertensive (Rana *et al.*, 2021), anticonvulsive (Chopra *et al.*, 2020) and antibacterial activities (Chopra and Sahu, 2020). Imidazole exhibits as anthelmintics and nematocides stimulated considerable interest in exploring the possible synthesis of potential condensed heterocyclic systems in which biologically active imidazole ring is fused with quinoxaline moieties. A large number of quinoxaline derivatives have been reported as bactericides, fungicides, herbicides and insecticides, 2-p-chloro-phenyl-3-chloroquinaxoline useful as fungicide against bean powdery mildew, bean rust, riceblast and angular leaf spot of cucumber. Some 2-styryl-4-amino-6-methoxyquinaxolines have been claimed as bactericides which also inhibited the growth of fungi at higher concentration. 4-Diethylamino-2-p-chlorostyryl-7-chloroquinaxaline was found to be effective bactericides against *Mycobacterium tuberculosis* at 0.25-0.5 µg/ml. In view of above, to synthesize the compound (2) with the anticipation that a combination of 1,3,4-oxadiazole moiety with quinoxaline structure might result in compounds of enhanced fungicidal activity.



(1)



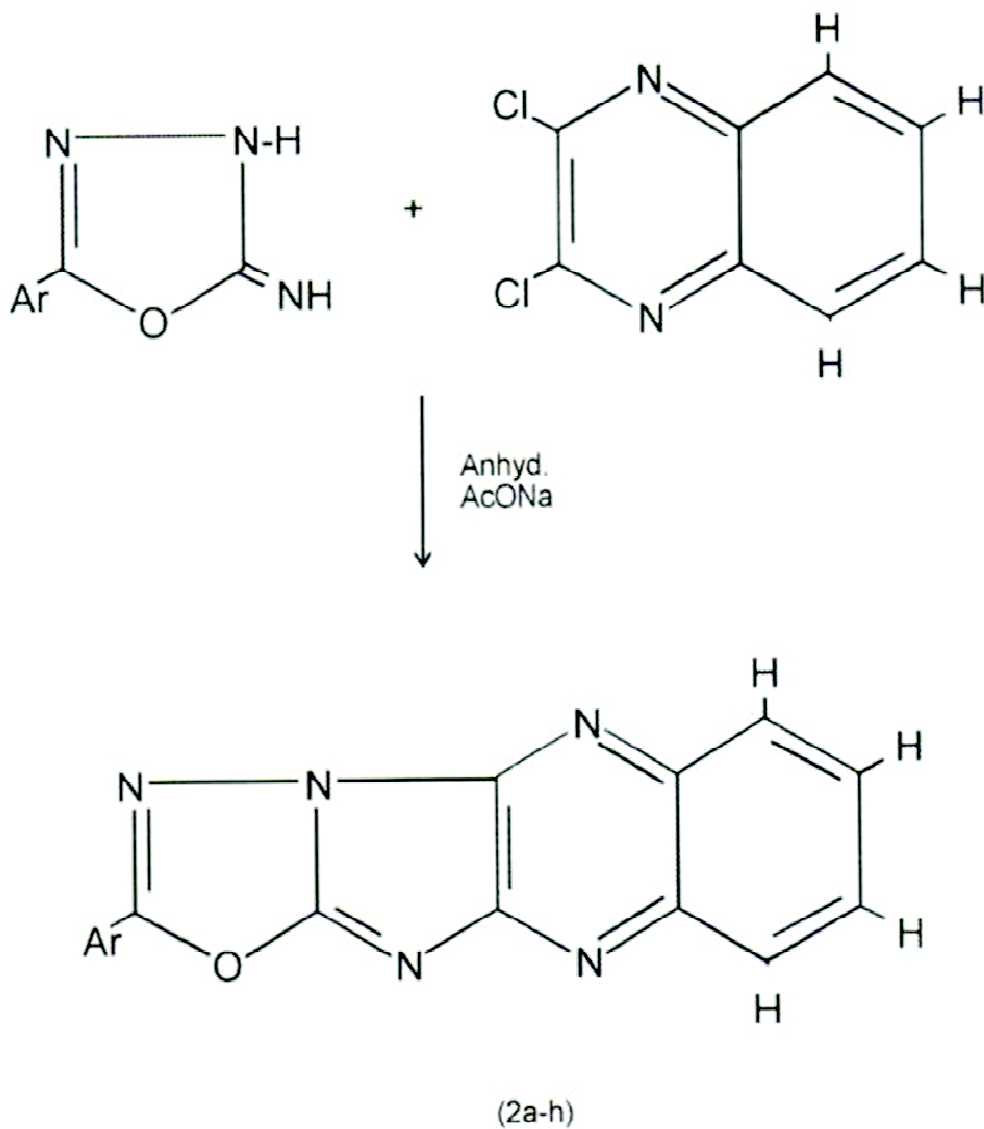
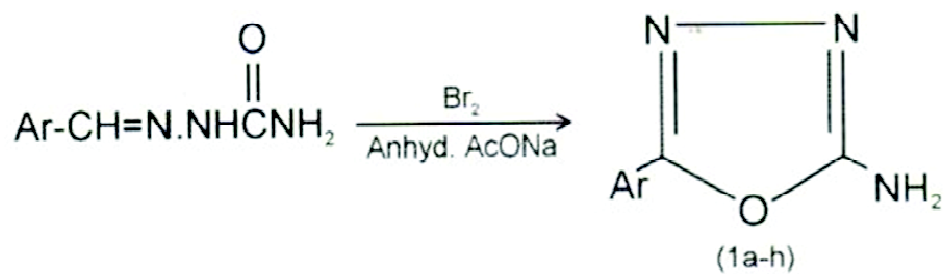
(2)

EXPERIMENTAL

2-aryl 1,3,4-oxadiazolo [3,2-6] imidazo [4,5-5] quinoxalines were prepared by the method of Mohan and Kumar, 2003. A solution of 2-amino-5-aryl-1,3,4-oxadiazole (1.169, 0.0005 mole), 2,3-dichloroquinaxaline (0.999, 0.005 mole) and anhydrous sodium acetate (0.829, 0.01 mole) in absolute ethanol (70 ml.) was heated under reflux for 6 hours. The reaction mixture was concentrated, cooled and poured into cold water. A ppt. obtained which was filtered, dried & recrystallized from methanol. The other compounds prepared in this way have been listed in Table 1.

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CHEMICAL REACTIONS



For 1,2; Ar = a, Phenyl; b, 4-Nitrophenyl; c, 2-Nitrophenyl; d, 4-Chlorophenyl; e, 2-Chlorophenyl; f, 4-Methylphenyl; g, 3-Methylphenyl.

RESULTS AND DISCUSSION

The physical data and spectral data of some compounds are recorded in Table 2.

Table 1: 2-ARYL 1,3,4- OXADIAZOLO [3,2-6] IMIDAZO [4,5-5] QUINOXALINES

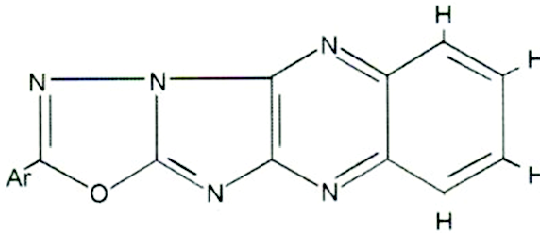
						
(1 a-h)						
Co.No.	Ar	M.P. °C	Yield %	M.F.	Analysis N%	
					Found	Calculated
1a	C ₆ H ₅	195	61	C ₁₆ H ₉ N ₅ O	24.32	24.39
1b	p-NO ₂ -C ₆ H ₄	262	57	C ₁₆ H ₈ N ₆ O ₃	25.24	25.30
1c	o-NO ₂ -C ₆ H ₄	264	55	C ₁₆ H ₈ N ₆ O ₃	25.20	25.30
1d	p-Cl-C ₆ H ₄	236	50.36	C ₁₆ H ₈ N ₅ OCl	21.69	21.77
1e	o-Cl-C ₆ H ₄	238-240	52	C ₁₆ H ₈ N ₅ OCl	21.67	21.77
1f	p-CH ₃ -C ₆ H ₄	--	52	C ₁₇ H ₁₁ N ₅ O	23.19	23.25
1g	o-CH ₃ -C ₆ H ₄	258	54	C ₁₇ H ₁₁ N ₅ O	23.17	23.25
1h	m-CH ₃ -C ₆ H ₄	267	56.02	C ₁₇ H ₁₁ N ₅ O	23.19	23.25

Table 2: Spectral data of some representative number of compounds

Comp. No.	IR D _{max} C _m ⁻¹	¹ H NMR δ
1a-	1525 (C-N), 1600 (C=C) 1610 (C=N) 3040 (aromatic C-H stretching) 700, 740, 780, 860 (Disubstituted benzene ring)	7.29-8.12 (9H,m, aromatic proton)
1c-	1532 (C-N), 1605 (C=C) 1615 (C=N) 3052 (aromatic C-H stretching) 710, 760, 810, 840 (Substituted benzene nucleus)	7.07-8.15 (8H,m, aromatic proton)
1g-	1540 (C-N), 1610 (C=C) 1630 (C=N) 3065 (aromatic C-H stretching) 712, 765, 815, 860 (Substituted benzene nucleus)	2.49 (3H, S, CH ₃ proton) 7.25-8.20 (8H,m, aromatic proton)

Five such compounds have been screened for their antibacterial activity against two bacterial species. The screening data have been reported in Table-3.

It is observed from the antibacterial data that these compounds are moderately active against both organism at higher concentration.

Table 3: Number of Replication in Each Case = 3

Compound No.	Zone of Inhibition (m.m.)			
	S. aureus		E. coli	
	Concentrations used		Concentrations used	
	100 μgml^{-1}	10 μgml^{-1}	100 μgml^{-1}	10 μgml^{-1}
1a	12	11	11	10
1c	20	18	19	17
1e	18	15	16	13
2f	17	14	15	12
2g	15	12	13	10
Amphicillin	26	24	22	19

CONCLUSION

The screening data indicates that all the tested compounds are more active against both *S. aureus* and *E. coli*. The compound 1a and 1g are less active than the compound 1a,1c and 1f indicating that introduction of methyl group in aromatic nucleus reduces its activity. Further it is observed from the screening data that introduction of Nitro group or Chloro group in aryl moiety present at heterocyclic ring increase the antibacterial activity.

ACKNOWLEDGEMENT

Author is extremely thankful to Dr R.S. Singh (Ex Head & Ex Principal), Department of Chemistry, T.D.P.G. College, Janunpur for his valuable guidance.

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